Copper-based nanopesticides: synthesis and application of more sustainable plant protection products

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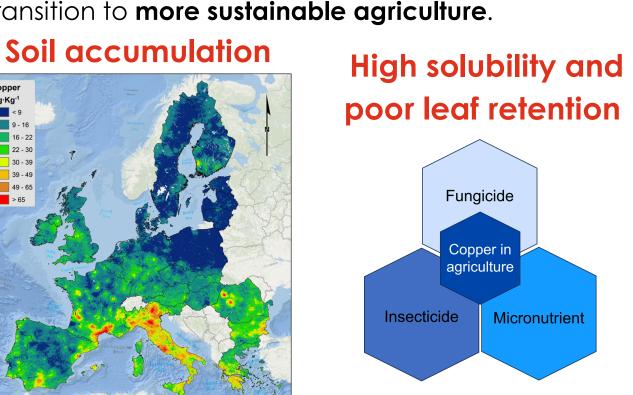
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Bioinspired nanocarrier for copper delivery

Copper-based compounds are widely applied in agriculture for their fungicidal and insecticidal properties, and as a micronutrient source. However, their intensive use and high soil accumulation of this metal have led to contamination in several cropping systems. In light of these environmental concerns, EU regulatory bodies have progressively restricted copper applications in order to foster more sustainable agricultural practices. Consequently, the development of innovative formulations with enhanced efficacy and reduced environmental impact can support the transition to more sustainable agriculture.



Experimental

Cu(II) Cu(I)

 $2p_{1/2}$ $2p_{1/2}$

Envelope

cit10

1 wt% Cu

cit40

Law Restrictions

Cu(II) Cu(I)

 $2p_{3/2}$ $2p_{3/2}$

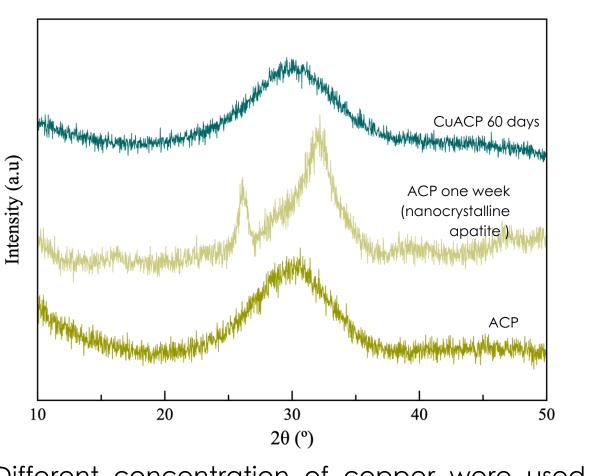
CuACP

pH-responsive solubility

ca²PO4²Cu²
Easily doped with metals

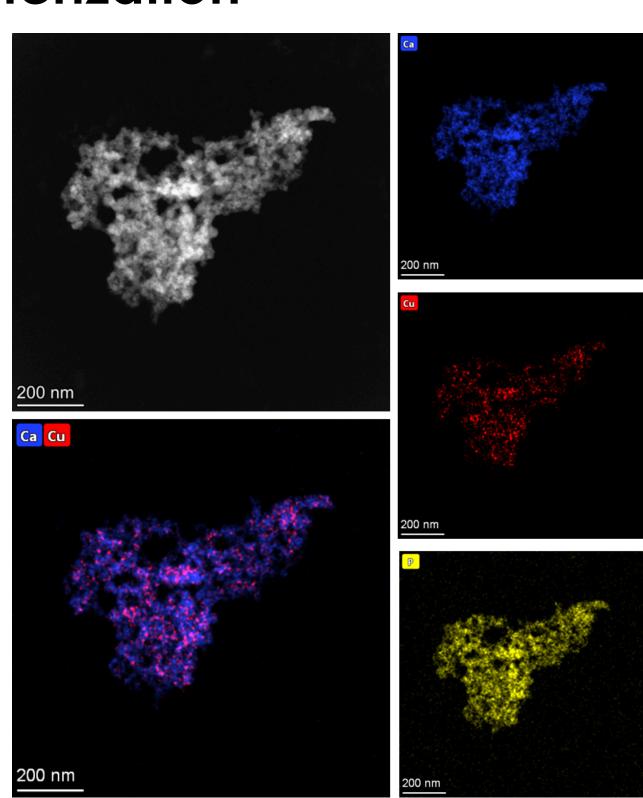
Amorphous calcium phosphate Ca₃(PO₄)₂_nH₂O) is precursor in bone hard tissue formation. It presents desireable properties for adsorption and delivery of ions and molecules due to its structure and pH dependent properties.4 ACP has shown promising results as agriculture, including in nanocarrier applications such as **biostimulants** based on methyl jasmonate⁵ or **biofortification** and protection through zinc delivery. With copper doping, the aim is to explore its effects in fungal control and nutrition.

Characterization



Different concentration of copper were used for the synthesis. **Powder X-ray diffraction** patterns of all CuACP confirmed the synthesis of the **amorphous** form. Moreover, copper stabilized the amorphous phase for more than 6 months, and no crystalline coprecipitation was observed.

TEM microrpahs and EDS elemental maps of CuACP showing homogeneous Cu (red) distribution within the nanoparticle, along with Ca (blue) and P (yellow) of ACP.



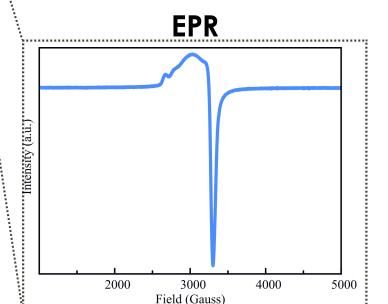
High-angle annular dark field-scanning transmission electron microscopy (HAADF-STEM) images and energy-dispersive X-ray (EDS) elemental maps

XPS analysis

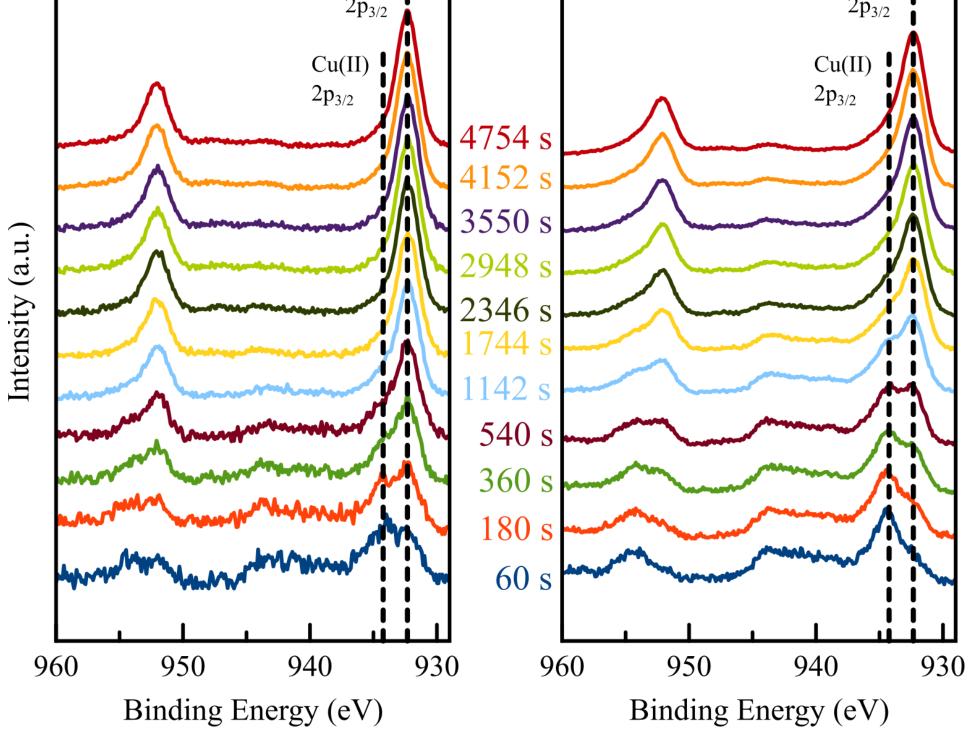
X-ray Photoelectron Spectroscopy (XPS) was employed to investigate the oxidation state of copper in the synthesized materials. In all samples copper a partial or even complete conversion of Cu^{2+} (used as reactant) to Cu^{+} was observed.

Several **hypotheses** were considered: i) reduction of copper by reactants, ii) precipitation of amorphous by-products, or iii) an artefact of the analysis (i.e., beam-induced surface reduction).

After eliminating potential reducing agents from the synthesis and still observing the same conversion, study of EPR evidenced the presence of Cu(II) in all samples. Thus, time-resolved XPS under controlled X-ray exposure was carried out to investigate **beam-induced reduction** hypotesis, confirming the progressive reduction of copper with increasing exposure.* Therefore, cit70 (9 % Cu) was selected for following experiments.



Electron Paramagnetic Resonance (EPR) of cit10. Since Cu(I) is EPR-silent, no signal was expected; however, a clear signal was detected, supporting the beam-induced reduction hypothesis.



Cu(I) ■

Time resolved XPS under controlled exposure times. On the left **cit40** sample and **cit70** on the right. Notably, in the sample with the lowest copper amount, the conversion was immediate, preventing any time-dependent monitoring.

Alternaria Solani

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CuACP

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CuACP

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CuACP

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copper h

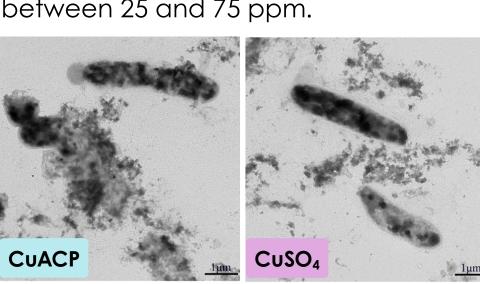
(KDOS),
significant
the patho

Alternaria solani demonstrated that, at 200 ppm of copper, CuACP provided a stronger inhibition of mycelial growth compared to the commercial copper hydroxide formulation (KDOS), which showed significantly different control of the pathogen.

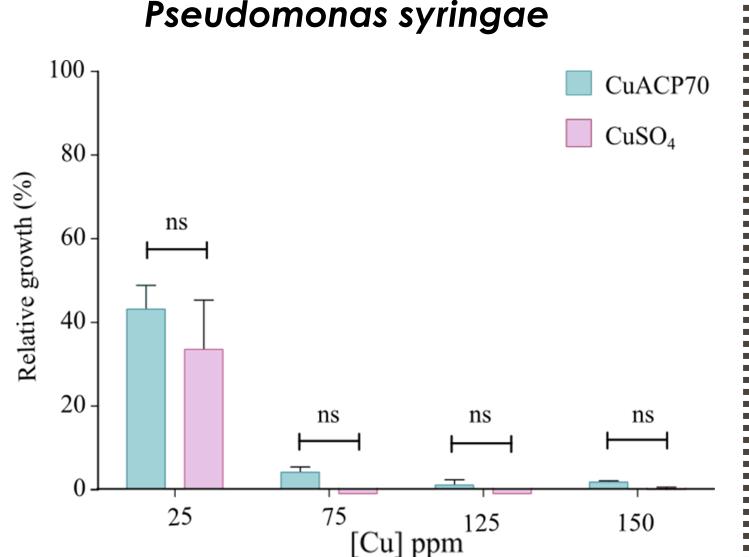
post hoc test when p < 0.05.

In the case of bacterium Pseudomonas

syringae, the antibacterial activity was evaluated across increasing copper concentrations and compared with copper sulfate. Both products displayed similar control of bacterial growth, with IC₅₀ values estimated between 25 and 75 ppm.



Bright-field TEM micrographs of bacterial suspension ofter 24 hours of contact with treatement.



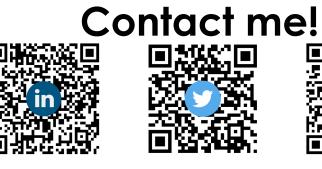
Antimicrobial activity was measured using MTS assay. Statistics: two-way ANOVA with Bonferroni post hoc test when p < 0.05.

SEM micrographs Leaf retention representative of the sample **Application** Rain simulation Analysis **CUACP** CuSO₄ CuACP commercial KDOS CuSO, retention was tomato tested and assessed with ICP-OES. The results showed that, after rain simulation, leaves treated with CuACP, retained 60% of the initially present, compared to copper commercial CuSO₄ and KDOS, which retained 30% and 24%, respectively.

Conclusions

This work presents promising results for a new copper-based nanomaterial. Solid-state analysis confirmed the presence of 9% wt Cu(II) in the amorphous nanomaterial, while antimicrobial tests show increased in vitro efficacy of CuACP against A. Solani and equivalent control of P. syringae, compared to conventional products. Planned greenhouse experiments will soon be conducted to integrate in vitro and leaf retention results, which so far suggest enhanced performance for CuACP. Finally, soil ecotoxicity studies will be carried out to assess the environmental impact of CuACP nanoparticles.





I will be happy to hear your thoughts